# **Combining Ring-Closing Metathesis and Hydroformylation** Strategies: A Novel Approach to Spirocyclic y-Butyrolactones

Bernd Schmidt,\* Burkhard Costisella, Rafael Roggenbuck, Markus Westhus, Holger Wildemann, and Peter Eilbracht

Universität Dortmund, Fachbereich Chemie, Organische Chemie, Otto-Hahn-Strasse 6, D-44227 Dortmund, Germany

bschmidt@chemie.uni-dortmund.de

Received May 19, 2001

Di- or tetrahydropyrans with a vinyl side chain are obtained by diastereoselective ring-closing metathesis or by addition of vinylmagnesium chloride to an appropriately functionalized tetrahydropyranone. The resulting allylic alcohols are converted to spirocyclic hemiacetals under hydroformylation conditions. Oxidation yields the corresponding lactones.

#### Introduction

Spirocyclic  $\gamma$ -butyrolactones not only are widespread in nature<sup>1-4</sup> but have also played a key role as synthetic intermediates.<sup>5–7</sup> One example of a natural product with a spirocyclic  $\gamma$ -butyrolactone moiety is the alkaloid stemotinine,<sup>8</sup> and the antitumor antibiotic aranorosin may serve as an example where a spirocyclic lactone has been used as a key intermediate in the total synthesis<sup>9</sup> (Chart 1). Furthermore, a variety of methods are available for the conversion of  $\gamma$ -butyrolactones into  $\alpha$ -methylene- $\gamma$ butyrolactones.<sup>10</sup> Spirocyclic, steroidal derivatives of these compounds often show outstanding biological properties.<sup>3,4</sup> In principle, one can imagine two different strategies for the construction of spiro compounds: (i) the conversion of a tetrafunctional acyclic compound into a spirocyclic system in a one-pot reaction or (ii) the stepwise formation of two carba- or heterocycles via a monocyclic intermediate. A variety of methods for the synthesis of lactones is available, for example, lactonization after asymmetric dihydroxylation, iodolactonization, palladiumcatalyzed cyclization of allenic or homopropargylic carboxylic acids, or samarium-mediated addition of ketones to acrylates. This subject has been reviewed several times over the past few years.<sup>11–14</sup>

\* To whom correspondence should be addressed. Fax: +49 (231) 7555363.

- (1) Nagahisa, M.; Koike, K.; Narita, M.; Ohmoto, T. Tetrahedron **1994**, *50*, 10859–10866.
- (2) Koike, K.; Suzuki, Y.; Ohmoto, T. Phytochemistry 1994, 35, 701-704.
- (3) Sawant, M. S.; Katoch, R.; Trivedi, G. K.; Desai, U. R. J. Chem. Soc., Perkin Trans. 1 1998, 843-846.
- (4) Sawant, M. S.; Nadkarni, P. J.; Desai, U. R.; Katoch, R.; Korde, S. S.; Trivedi, G. K. *J. Chem. Soc., Perkin Trans.* 1 1999, 2537–2542.
  (5) Marschall, H.; Penninger, J.; Weyerstahl, P. *Liebigs Ann. Chem.*
- 1982. 49-67 (6) Marschall, H.; Penninger, J.; Weyerstahl, P. Liebigs Ann. Chem.
- 1982, 68-72. (7) Reid, A. M.; Steel, P. G. J. Chem. Soc., Perkin Trans. 1 1998, 2795-2801.
- (8) Xu, R.-S.; Lu, Y.-J.; Chu, J.-H.; Iwashita, T.; Naoki, H.; Naya,
- (a) Na, N. B., Ed. 1.5., Chi, S. Ti., Ivashita, T., Ivashi, H., Ivashi, Y.; Nakanishi, K. *Tetrahedron* **1982**, *38*, 2667–2670.
  (9) Wipf, P.; Kim, Y.; Fritch, P. C. *J. Org. Chem.* **1993**, *58*, 7195– 7203.
- (10) Hoffmann, H. M. R.; Rabe, J. Angew. Chem. 1985, 97, 96-112.
- (11) Collins, I. Contemp. Org. Synth. **1996**, *3*, 295–321. (12) Collins, I. Contemp. Org. Synth. **1997**, *4*, 281–307.
- (13) Collins, I. J. Chem. Soc., Perkin Trans. 1 1998, 1869–1888.
- (14) Collins, I. J. Chem. Soc., Perkin Trans. 1 1999, 1377-1395.



Aranorosin

In this contribution, we present a novel concept for the synthesis of spirocyclic lactones, which is based on a ruthenium-catalyzed olefin metathesis and a rhodiumcatalyzed hydroformylation-acetalation sequence (Scheme 1).

We demonstrate the utility of our approach for the example of di- or tetrahydropyrans linked to lactones in a spirocyclic fashion. Although this class of substances might be interesting in itself,<sup>5,6</sup> they may also serve, after cleavage of the lactone, as intermediates in the synthesis of pyran derivatives with functionalized side chains. This structural element is found in several natural products such as the pseudomonic acids.<sup>15</sup>

The ring-closing olefin metathesis<sup>16–18</sup> has become one of the most powerful cyclization reactions since the

(18) Fürstner, A. Angew. Chem. 2000, 112, 3140-3172.

 <sup>(15)</sup> Class, Y. J.; DeShong, P. Chem. Rev. 1995, 95, 1843–1857.
 (16) Armstrong, S. K. J. Chem. Soc., Perkin Trans. 1 1998, 371–

<sup>388</sup> 

<sup>(17)</sup> Schuster, M.; Blechert, S. Angew. Chem., Int. Ed. Engl. 1997, 36 2036-2055



introduction of highly efficient precatalysts by Schrock<sup>19</sup> and Grubbs.<sup>20</sup> The RCM reaction has been used in the synthesis of spirocycles, either for cyclization of two rings by double-ring-closing metathesis<sup>21-27</sup> or by single-ringclosing metathesis starting from an appropriately functionalized monocyclic precursor.28-33

Hydroformylation of olefins not only is a well-established method for the industrial production of aldehydes but has also found interesting applications in organic synthesis.<sup>34</sup> The incorporation of a hydroformylation step in sequential reactions has proven to be a particularly valuable concept.<sup>35</sup> For example, the hydroformylation of an allylic or homoallylic alcohol yields hemiacetals via formation of the primarily formed aldehydes.<sup>36–38</sup> The hemiacetals may be converted into cyclic enol ethers by elimination<sup>39</sup> or into lactones by oxidation.<sup>40</sup>

- (19) Schrock, R. R.; Murdzek, J. S.; Bazan, G. C.; Robbins, J.; (20) Fu. G. C.; Nguyen, S. T.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. 1990, 112, 3875–3886.
   (20) Fu. G. C.; Nguyen, S. T.; Grubbs, R. H.; Ziller, J. W. J. Am. Chem. Soc. 1993, 115, 9856–9857.
- (21) Bassindale, M. J.; Hamley, P.; Leitner, A.; Harrity, J. P. A. *Tetrahedron Lett.* **1999**, *40*, 3247–3250.
- (22) Wallace, D. J.; Cowden, C. J.; Kennedy, D. J.; Ashwood, M. S.; Cottrell, I. F.; Dolling, U.-H. *Tetrahedron Lett.* **2000**, *41*, 2027–2029.
- (23) Wallace, D. J.; Bulger, P. G.; Kennedy, D. J.; Ashwood, M. S.; Cottrell, I. F.; Dolling, U.-H. Synlett **2001**, 357–360.
- (24) Wallace, D. J.; Goodman, J. M.; Kennedy, D. J.; Davies, A. J.; Cowden, C. J.; Ashwood, M. S.; Cottrell, I. F.; Dolling, U.-H.; Reider, P. J. Org. Lett. 2001, 3, 671-674.
- (25) Schmidt, B.; Westhus, M. *Tetrahedron* **2000**, *56*, 2421–2426. (26) Schmidt, B.; Wildemann, H. J. Chem. Soc., Perkin Trans. 1 2000. 2916-2925
- (27) Schmidt, B.; Wildemann, H. J. Org. Chem. 2000, 65, 5817-5822
- (28) Evans, P.; Grigg, R.; Ramzan, M. I.; Sridharan, V.; York, M. Tetrahedron Lett. 1999, 40, 3021-3024.
- (29) van Hooft, P. A. V.; Leeuwenburgh, M. A.; Overkleeft, H. S.; van der Marel, G. A.; van Boeckel, Č. A. A.; van Boom, J. H. *Tetrahedron Lett.* **1998**, *39*, 6061–6064.
- (30) van Hooft, P. A. V.; van der Marel, G. A.; van Boeckel, C. A. A.; van Boom, J. H. Tetrahedron Lett. 2001, 42, 1769-1772
- (31) Holt, D. J.; Barker, W. D.; Jenkins, P. R.; Davies, D. L.; Garratt, S.; Fawcett, J.; Russell, D. R.; Ghosh, S. Angew. Chem. 1998, 110, 3486-
- 3488 (32) Holt, D. J.; Barker, W. D.; Jenkins, P. R.; Panda, J.; Ghosh, S.
- *J. Org. Chem.* **2000**, *65*, 482–493. (33) Grigg, R.; Sridharan, V.; York, M. *Tetrahedron Lett.* **1998**, *39*, 4139 - 4142
- (34) Breit, B.; Seiche, W. Synthesis 2001, 1-36.
- (35) Eilbracht, P.; Bärfacker, L.; Buss, C.; Hollmann, C.; Kitsos-Rzychon, B. E.; Kranemann, C. L.; Rische, T.; Roggenbuck, R.; Schmidt,
- A. Chem. Rev. 1999, 99, 3329-3365. (36) Anastasiou, D.; Jackson, W. R. Aust. J. Chem. 1992, 45, 21-
- 37 (37) Trzeciak, A. M.; Wolszczak, E.; Ziólkowski, J. J. New. J. Chem. 1996, 20, 365-370.
- (38) Kitsos-Rzychon, B.; Eilbracht, P. Tetrahedron 1998, 54, 10721-10732
- (39) Roggenbuck, R.; Eilbracht, P. Tetrahedron Lett. 1999, 40, 7466-7456.
- (40) Breit, B. Angew. Chem. 1996, 108, 3021-3023.

## **Results and Discussion**

Preparation of Monocyclic Precursors. Starting from the protected allyl-homoallyl ether 2 (obtained by benzylation of alcohols 1),<sup>26</sup> we obtained the dihydropyrans 3 by ring-closing metathesis. Epoxidation was moderately diastereoselective for the methyl derivative **3a** (diastereomeric ratio = 2:1), whereas the phenyl derivative 3b was obtained as a 6:1 mixture of diastereoisomers. Compound 4a was treated as a mixture of diastereomers with benzylic alcohol in the presence of borontrifluoride etherate. Upon chromatography, the major isomer  $(3S^*, 4S^*, 6R^*)$ -**5a** was isolated, which was used for further transformations. In the case of epoxide 4b, the minor diastereomer was removed by flash chromatography. Epoxide cleavage under the same conditions gave (3*S*\*,4*S*\*,6*S*\*)-**5b** and (2*S*\*,4*R*\*,5*R*\*)-**5b** as a 4.5:1 mixture of regioisomers. Oxidation of the hydroxyl functionality was achieved using TPAP/NMO<sup>41</sup> to yield the ketones 6a,b. Attack of vinylmagnesium chloride occurred from the face opposite to the benzyloxy substituent to give the tertiary allylic alcohols 7a,b with high diastereoselectivity (only one diastereomer was detected in the <sup>1</sup>H NMR spectrum) (Scheme 2).

We have recently devised an alternative synthesis for pyran derivatives with an exo-vinyl moiety. Ring-closing metathesis of divinyl carbinols 8a and 8b gives dihydropyrans **9a** and **9b** in diastereomeric ratios of 3:1 and 4:1, respectively.<sup>27,42</sup> Ring-closing metathesis of triene 10 yields dihydropyrans (5*R*\*,8*S*\*)-11 and (5*S*\*,8*S*\*)-11 as an easily separable mixture of diastereomers (Scheme 3).

Preparation of Spirocyclic Products. Allylic alcohols 7a,b were subjected to hydroformylation in the presence of Rh(acac)(CO)<sub>2</sub> (1.0 mol %) and BIPHEPHOS (4.0 mol %)<sup>43</sup> under 10 bar of carbon monoxide and 10 bar of hydrogen at 60 °C. Under these mild conditions, hemiacetals 12a,b result as mixtures of anomers, which is indicated by signals at approximately 99 ppm in the carbon NMR spectrum. Side products resulting from formation of the branched aldehyde could not be detected. The crude hemiacetals were oxidized to the lactones **13a**,**b** using TPAP/NMO. Compound (5*R*\*,8*R*\*,10*S*\*)-**13a** was obtained from 7a (single isomer), and 7b (mixture of regioisomers) was converted to (5R\*,8R\*,10S\*)-13b and  $(5S^*, 6R^*, 9S^*)$ -13b. The regioisomers were separated by column chromatography and fully characterized (Scheme 4).

Hydroformylation of dihydropyrans 9a, 9b, (5R\*,8S\*)-11, and (5*S*<sup>\*</sup>,8*S*<sup>\*</sup>)-11 is highly regio- and chemoselective, with only the exo-vinyl moiety being attacked under the reaction conditions. The hemiacetals 12, 14, and 16 were obtained as inseparable mixtures of anomers, which were immediately oxidized to the corresponding lactones 13, 15, and 17 using the TPAP/NMO reagent combination. Compounds 9a,b were employed as 3:1 and 4:1 mixtures of diastereoisomers. The resulting lactones (5S,6S)- and (5R,6S)-15a and  $(5S^*,6S^*)$ - and  $(5R^*,6S^*)$ -15b could be easily separated by flash chromatography on silica. Compounds (3R\*,6S\*)-11 and (3S\*,6S\*)-11 were employed as single diastereomers. In these cases, single isomers of the corresponding lactones  $(5R^*, 8S^*)$ -17 and  $(5S^*, 8S^*)$ -17 were obtained (Scheme 5).

<sup>(41)</sup> Griffith, W. P.; Ley, S. V. Aldrichimica Acta 1990, 23, 13-19.

<sup>(42)</sup> Schmidt, B.; Wildemann, H. Synlett 1999, 1591-1593

<sup>(43)</sup> Cuny, G. D.; Buchwald, S. L. J. Am. Chem. Soc. 1993, 115, 2066 - 2068



<sup>a</sup> Key: (i) NaH, BnBr, THF (90% of **2a**, 91% of **2b**); (ii) Cl<sub>2</sub>(Cy<sub>3</sub>P)<sub>2</sub>Ru=CHPh (1 mol %), DCM (81% of **3a**, 88% of **2b**); (iii) MCPBA, DCM (91% of **4a**, 99% of **4b**); (iv) BnOH, BF<sub>3</sub>OEt<sub>2</sub> (10 mol %), DCM, 0 °C, chromatography (57% of (3*S*\*,4*S*\*,6*R*\*)-**5a**, 78% of (3*S*\*,4*S*\*,6*S*\*)-**5b** and (2*S*\*,4*R*\*,5*R*\*)-**5b**) (4.5:1); (v) TPAP (5 mol %), NMO, 4 Å molecular sieve, DCM (84% of **6a**, 82% of **6b**); (vi) H<sub>2</sub>C=CHMgCl, ether, −78 °C (66% of **7a**, 89% of **7b**).



NOE Experiments. The relative configuration of all spirocyclic products was investigated by one- or twodimensional NOE experiments conducted at 600 or 500 MHz, respectively. For spirocyclic tetrahydropyran 13a, NOE interactions between the methyl group in the 2-position and the proton H3ax and between the  $-CH_2$ -OBn moiety and the proton H6ax indicate that the molecule preferably adopts the conformation depicted in Scheme 6. One proton of the CH<sub>2</sub> group of the lactone moiety shows NOE interactions with both protons H6, whereas the other proton of this CH<sub>2</sub> group shows a NOE interaction with H4. The phenyl analogue  $(5R^*, 8R^*, -$ 10S\*)-13b shows similar NOE interactions. Its regioisomer  $(5S^*, 6R^*, 9S^*)$ -**13b** adopts a conformation with the phenyl substituent in an axial position: NOE interactions between the ortho-H of the phenyl moiety and H6ax as well as H3eq are observed. A NOE interaction between the axially oriented H5 and one proton of the methylene moiety of the lactone ring is indicative of the relative configuration shown in Scheme 6.

NOE interactions for spirocyclic dihydropyrans **15** are summarized in Scheme 7. A NOE interaction between H6ax and H2 indicates that the phenyl moiety adopts a pseudoequatorial position. In the case of  $(5.S^*, 6.S^*)$ -**15b**, a NOE interaction between H2 and one proton of the methylene group of the lactone moiety is observed, which is missing for the other diastereomer. For the  $(5R^*, 6.S^*)$ isomer, a NOE interaction between the ortho protons of the phenyl group and one proton of the methylene group is indicative for the proposed relative configuration. For the methyl derivative **15a**, analogous NOE interactions were observed for the protons of the methyl group.

For dihydropyrans **17** with the spirocyclic junction in the 5-position, NOE interactions of the protons H6ax and H6eq with the methylene group of the lactone ring are most indicative of the relative configuration. For both diastereomers, a NOE effect is observed between the singlet of H2 and one doublet of the H6-AB-system; assignment of the corresponding signals to H6ax and H6eq, respectively, is made on the basis of this interaction. In ( $5S^*$ , $8S^*$ )-**17**, a NOE interaction between H6ax and one proton of the methylene group of the lactone is observed, whereas in ( $5R^*$ , $8S^*$ )-**17**, H6eq interacts with this CH<sub>2</sub> group (Scheme 8).

In conclusion, we have developed a synthetic approach toward bicyclic spiro compounds with a lactone moiety linked to a six-membered oxacycle on the basis of ringclosing metathesis and a hydroformylation-acetalation sequence. Use of the BIPHEPHOS ligand system allows the differentiation between exocyclic and endocyclic C-C



(5R\*, 8R\*, 10S\*)-13a

(3R\*, 4S\*, 6R\*)-7a + (2S\*, 4R\*, 5R\*)-7b (4.5 : 1)



(5R\*, 8R\*, 10S\*)-12b (5S\*, 6R\*, 9S\*)-12b



(5R\*, 8R\*, 10S\*)-13b (5S\*, 6R\*, 9S\*)-13b

<sup>a</sup> Key: (i) Rh(acac)(CO)<sub>2</sub> (1.0 mol %), BIPHEPHOS (4.0 mol %), CO (10 bar), H<sub>2</sub> (10 bar), dioxane; (ii) TPAP (5 mol %), NMO, 4 Å molecular sieve, DCM (80% of **13a**, 54% of **13b**).

double bonds under hydroformylation conditions, making spirocyclic dihydropyrans accessible. Further synthetic applications for the combination of olefin metathesis and hydroformylation strategies are currently under investigation in our laboratory.

### **Experimental Section**

Instrumentation, product identification, and general experimental methods have been described previously.<sup>44</sup> Hydroformylation reactions were conducted in steel pressure vessels with a PTFE insert. Signal assignment for NMR spectra is made on the basis of 2D methods (H, H–COSY) or NOE experiments. Signal assignment for cyclic products follows a numbering scheme where the oxygen atom is numbered 1 and the  $\alpha$ -carbon atom bearing a substituent is C2. The number of coupled protons was analyzed by DEPT experiments and is denoted by a number in parantheses following the  $\delta_{\rm C}$  value. One-dimensional, gradient-selected NOE spectra were obtained at 600 MHz with a mixing time of 800 ms. Two-

dimensional NOESY spectra were obtained at 500 MHz with a mixing time of 1.5 s.

2-Allyloxy-2-methyl-pent-4-enyloxymethylbenzene (2a). To a solution of the primary alcohol 1a (8.65 g, 55 mmol) in THF (150 mL) was added NaH (60% dispersion in mineral oil, 6.65 g, 166 mmol). The mixture was heated to reflux for 30 min and cooled to ambient temperature. Benzyl bromide (9.9 mL, 83 mmol) was added, and the mixture was again refluxed until the starting material was fully consumed as indicated by TLC. After aqueous workup, the solvent was evaporated and the residue was distilled (bp 130 °C, 0.4 mbar) to give 12.23 g (90%) of the benzyl ether 2a. IR (neat): 698 s, 1103 s, 1454 m, 2859 m, 2982 m. MS m/z (relative (rel) intensity): 247  $(M^+ + 1, <5)$ , 131 (60), 91 (100). <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  7.27–7.17 (5H), 5.83 (dddd, 1H, J = 17.1, 10.5, 5.3, 5.3 Hz), 5.73 (dddd, 1H, J = 17.5, 10.5, 7.0, 7.0 Hz), 5.18 (dddd, 1H, J = 17.1, 1.5, 1.5, 1.5 Hz), 5.03 (dddd, 1H, J = 10.5, 1.5, 1.5, 1.5 Hz), 4.98 (dddd, 1H, J = 17.5, 1.5, 1.5, 1.5 Hz), 4.94 (dddd, 1H, J = 10.5, 1.5, 1.5, 1.5 Hz), 4.45 (s, 2H), 3.93 (m, 1H), 3.89 (dddd, 1H, J = 11.8, 5.3, 1.5, 1.5 Hz), 3.32 (d, 1H, J = 9.8 Hz), 3.25 (d, 1H, J = 9.8 Hz), 2.30 (dddd, 1H, J = 14.8, 7.0, 1.5, 1.5 Hz), 2.27 (dddd, 1H, J = 14.8, 7.0, 1.5, 1.5 Hz), 1.10 (s, 3H). <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ 138.2 (0), 135.8 (1), 134.0 (1), 128.2 (1), 127.5 (1), 127.4 (1), 117.5 (2), 115.6 (2), 76.6 (0), 74.5 (2), 73.2 (2), 63.2 (2), 40.4 (2), 20.9 (3). Anal. Calcd for C<sub>16</sub>H<sub>22</sub>O<sub>2</sub>: C, 78.0; H, 9.0. Found: C, 77.6; H, 8.8.

2-Allyloxy-2-phenyl-pent-4-enyloxymethylbenzene (2b). Following the procedure for the methyl analogue, we obtained **2b** (16.0 g, 91%) from primary alcohol **1b** (12.5 g, 57 mmol) as a colorless liquid (bp 130 °C, 0.4 mbar). IR (neat): 699 s, 1072 s, 1103 s, 1496 m, 3075 m. MS *m/z* (rel intensity): 308 (M<sup>+</sup> -1, <5), 267 (5), 187 (25), 145 (25), 131 (80), 105 (80), 91 (100), 77 (25). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.32–7.13 (10H), 5.82 (dddd, 1H, J = 17.3, 10.3, 5.3, 5.3 Hz), 5.63 (dddd, 1H, J = 17.3, 10.3, 7.0, 7.0 Hz), 5.20 (dddd, 1H, J = 17.3, 1.5, 1.5, 1.5 Hz), 5.04 (dddd, 1H, J = 10.3, 1.5, 1.5, 1.5 Hz), 4.97 (dm, 1H, J = 17.3 Hz), 4.94 (dm, 1H, J = 10.3 Hz), 4.42 (d, 1H, J =12.8 Hz), 4.39 (d, 1H, J = 12.8 Hz), 3.76 (dddd, 1H, J = 12.8, 5.3, 1.5, 1.5 Hz), 3.73 (m, 1H), 3.70 (d, 1H, J = 9.5 Hz), 3.59 (d, 1H, J = 9.5 Hz), 2.75 (ddm, 1H, J = 14.3, 7.0 Hz), 2.64 (ddm, 1H, J = 14.3, 7.0 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$ 142.2 (0), 138.2 (0), 135.2 (1), 133.5 (1), 128.2 (1), 128.0 (1), 127.5 (1), 127.5 (1), 127.1 (1), 126.6 (1), 117.9 (2), 115.7 (2), 80.3 (0), 73.3 (2), 73.3 (2), 63.8 (2), 40.3 (2). Anal. Calcd for C<sub>21</sub>H<sub>24</sub>O<sub>2</sub>: C, 81.8; H, 7.8. Found: C, 82.1; H, 7.8.

2-Benzyloxymethyl-2-methyl-3,6-dihydro-2H-pyran (3a). To a solution of 2a (5.00 g, 20 mmol) in DCM (50 mL) was added Grubbs' catalyst (167 mg, 1 mol %). The reaction mixture was stirred at ambient temperature until the starting material was fully consumed as indicated by TLC. The solvent was evaporated, and the residue was purified by flash chromatography on silica using 5:1 (v/v) cyclohexane/MTBE as the eluent to give the dihydropyran 3a (3.60 g, 81%). IR (neat): 746 m, 1091 s, 1454 m, 2829 m, 3033 m. MS m/z (rel intensity): 219 ( $M^+$  + 1, 40), 201 (30), 181 (60), 153 (70), 129 (100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.27-7.16 (5H), 5.67 (dm, 1H, J = 10.3 Hz), 5.61 (dm, 1H, J = 10.3 Hz), 4.53 (d, 1H, J= 12.5 Hz), 4.48 (d, 1H, J = 12.5 Hz), 4.11 (dm, 1H, J = 16.8Hz), 4.06 (dm, 1H, J = 16.8 Hz), 3.33 (d, 1H, J = 9.8 Hz), 3.24 (d, 1H, J = 9.8 Hz), 2.13 (dm, 1H, J = 17.5 Hz), 1.77 (dm, 1H, J = 17.5 Hz), 1.18 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  138.3 (0), 128.2 (1), 127.5 (1), 127.4 (1), 125.1 (1), 122.6 (1), 75.7 (2), 73.4 (2), 71.7 (0), 61.0 (2), 31.7 (2), 20.9 (3). Anal. Calcd for C14H18O2: C, 77.0; H, 8.3. Found: C, 76.0; H, 8.1.

<sup>(44)</sup> Schmidt, B. J. Chem. Soc., Perkin Trans. 1 1999, 2627-2637.

### Schmidt et al.



<sup>*a*</sup> Key: (i) Rh(acac)(CO)<sub>2</sub> (1.0 mol %), BIPHEPHOS (4.0 mol %), CO (10 bar), hydrogen (10 bar), dioxane; (ii) TPAP (5 mol %), NMO, 4 Å molecular sieve, DCM (52-81% over two steps).



(d, 1H, J = 10.0 Hz), 3.37 (d, 1H, J = 10.0 Hz), 2.70 (dddd, 1H, J = 17.8, 2.5, 2.5, 2.5, 2.5 Hz), 2.55 (dddd, 1H, J = 17.8, 2.5, 2.5, 2.5, 2.5, 2.5 Hz), <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  140.9 (0), 138.2 (0), 128.2 (1), 128.1 (1), 127.5 (1), 127.4 (1), 127.3 (1), 126.8 (1), 125.7 (1), 122.8 (1), 76.3 (0), 77.1 (2), 73.4 (2), 61.7 (2), 28.3 (2). Anal. Calcd for C<sub>19</sub>H<sub>20</sub>O<sub>4</sub>: C, 81.4; H, 7.2. Found: C, 81.1; H, 7.2.





(5S\*, 6S\*)-15b (5R\*, 6S\*)-15b



(5S\*, 8S\*)-17 (5R\*, 8S\*)-17

4-Benzyloxymethyl-4-methyl-3,7-dioxabicyclo[4.1.0]heptane (4a). To a solution of the dihydropyran 3a (2.65 g, 12.1 mmol) in DCM (50 mL) was added MCPBA (70% dispersion in water, 3.14 g, 18.2 mmol). The mixture was stirred until the starting material was completely consumed as indicated by TLC. The reaction mixture was diluted with MTBE and washed with Na<sub>2</sub>SO<sub>3</sub> solution and then Na<sub>2</sub>CO<sub>3</sub> solution. The organic layer was dried, filtered, and evaporated, and the residue was purified by flash chromatography on silica to give epoxide **4a** (2.75 g, 97%) as a 2:1 mixture of diastereoisomers. Spectroscopic data for the major  $(1S^*, 4R^*, 6R^*)$ -isomer. IR (neat): 739 s, 1110 s, 1454 m, 2862 m. MS *m/z* (rel intensity):  $235 (M^+ + 1, 10), 181 (10), 143 (15), 113 (85), 91 (100), 65 (20).$ <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.34–7.22 (5H), 4.52 (d, 1H, J = 12.3 Hz), 4.45 (d, 1H, J = 12.3 Hz), 4.02 (d, 1H, J = 13.8Hz), 3.95 (d, 1H, J = 13.8 Hz), 3.32 (d, 1H, J = 9.5 Hz), 3.29 (dd, 1H, J = 9.8, 4.8 Hz), 3.20 (d, 1H, J = 9.5 Hz), 3.04 (m, 1H), 1.89 (d, 1H, J=15.3 Hz), 1.63 (dd, 1H, J=15.3, 5.7 Hz),

(1S\*,4S\*,6R\*)-4-Benzyloxymethyl-4-phenyl-3,7-dioxabicyclo[4.1.0]heptane (4b). Following the procedure for the methyl analogue, we obtained 4b (2.50 g, 99%) from 3b (2.39 g, 8.5 mmol) as a 6:1 mixture of diastereomers. The major isomer was separated by flash chromatography as a colorless solid (mp 69 °C). IR (KBr, disk): 699 m, 1104 s, 2858 m, 3028 m. MS  $\hat{m}/z$  (rel intensity): 297 (M<sup>+</sup> + 1, 10), 279 (15), 175 (100), 145 (20), 105 (30), 91 (40). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.38– 7.10 (10H), 4.49 (d, 1H, J = 12.5 Hz), 4.35 (d, 1H, J = 12.5Hz), 3.96 (d, 1H, J = 13.5 Hz), 3.61 (d, 1H, J = 13.5 Hz), 3.50 (d, 1H, J = 10.0 Hz), 3.40 (d, 1H, J = 6.5 Hz), 3.37 (d, 1H, J= 10.0 Hz), 2.79 (d, 1H, J = 4.3 Hz), 2.68 (dd, 1H, J = 16.1, 6.0 Hz), 2.37 (d, 1H, J = 16.1 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  139.4 (0), 138.1 (0), 128.5 (1), 128.2 (1), 127.7 (1), 127.5 (1), 127.4 (1), 127.4 (1), 77.7 (2), 75.2 (0), 73.3 (2), 59.4 (2), 48.6 (1), 48.5 (1), 27.6 (2). Anal. Calcd for  $C_{19}H_{20}O_3$ : C, 77.0; H, 6.8. Found: C, 76.8; H, 6.8.

(3S\*,4S\*,6R\*)-4-Benzyloxy-6-benzyloxymethyl-6-methyl-tetrahydropyran-3-ol (5a). To a solution of the epoxide 4a (2:1 mixture of diastereomers, 2.09 g, 8.9 mmol) and benzylic alcohol (9.3 mL, 89.4 mmol) was added borontrifluoride etherate (0.1 mL, 0.8 mmol) at 0 °C. The mixture was stirred at this temperature until the starting material was completely consumed as indicated by TLC and diluted with MTBE. The reaction was quenched with Na<sub>2</sub>CO<sub>3</sub> solution, and the mixture was dried with MgSO<sub>4</sub>, filtered, and evaporated. Upon flash chromatography on silica, only the  $(3S^*, 4S^*, 6R^*)$ isomer of 5a was isolated in pure form (1.73 g, 57%). IR (neat): 698 s, 737 s, 1094 s, 1454 m, 2867 m, 2929 m, 3436 br s. MS *m*/*z* (rel intensity): 341 (M<sup>+</sup> - 1, <5), 221 (40), 167 (20), 91 (100), 65 (20). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.36-7.26 (10H), 4.63 (d, 1H, J = 11.5 Hz), 4.57 (d, 1H, J = 12.3 Hz), 4.51 (d, 1H, J = 12.3 Hz), 4.44 (d, 1H, J = 11.5 Hz), 3.92 (dd, 1H, J = 11.8, 4.0 Hz), 3.62 (ddd, 1H, J = 7.5, 7.5, 4.0 Hz), 3.58 (ddd, 1H, J = 9.0, 7.5, 4.0 Hz), 3.48 (dd, 1H, J = 11.8, 7.5 Hz), 3.41 (d, 1H, J = 9.5 Hz), 3.36 (d, 1H, J = 9.5 Hz), 2.40 (bs, 1H), 2.21 (dd, 1H, J = 13.8, 4.0 Hz), 1.42 (dd, 1H, J = 13.8, 9.0 Hz), 1.29 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz): δ 138.2 (0), 138.1 (0), 128.5 (1), 128.4 (1), 127.8 (1), 127.7 (1), 127.6 (1), 127.5 (1), 77.2 (0), 77.0 (1), 73.7 (2), 73.5 (2), 70.8 (2), 69.5 (1), 64.0 (2), 34.8 (2), 25.0 (3). Anal. Calcd for C<sub>21</sub>H<sub>26</sub>O<sub>4</sub>: C, 73.7; H, 7.7. Found: C, 73.3; H, 7.4.

 $(3S^*, 4S^*, 6S^*) - 4 - Benzyloxy - 6 - benzyloxymethyl - 6 - phen$ yltetrahydropyran-3-ol (5b). Following the procedure for the methyl analogue, we obtained 5b (2.90 g, 78) from 4b (2.73 g, 9.2 mmol) as a 4.5:1 mixture of regioisomers. Spectroscopic data were obtained from the mixture. IR (neat): 699 s, 1075 s, 1096 s, 2866 m, 3426 br s. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$ 7.37 (d, 2H, J = 7.5 Hz), 7.27-7.10 (6H), 7.02 (d, 2H, J = 8.0 Hz), 4.49 (d, 1H, J = 11.8 Hz), 4.42 (d, 1H, J = 12.5 Hz), 4.35 (d, 1H, J = 11.8 Hz), 4.31 (d, 1H, J = 12.5 Hz), 3.97 (dd, 1H, J = 11.8, 3.5 Hz), 3.68 (ddd, 1H, J = 7.3, 6.3, 4.3 Hz), 3.62 (dd, 1H, J = 11.8, 6.0 Hz), 3.54 (ddd, 1H, J = 6.3, 6.0, 3.5 Hz), 3.48 (d, 1H, J = 10.0 Hz), 3.40 (d, 1H, J = 10.0 Hz), 2.76 (bs, 1H), 2.60 (dd, 1H, J = 14.3, 7.3 Hz), 1.92 (dd, 1H, J = 14.3, 4.3 Hz).  $^{13}{\rm C}$  NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  143.2 (0), 138.2 (0), 137.9 (0), 128.3 (1), 128.2 (1), 127.9 (1), 127.5 (1), 127.5 (1), 127.3 (1), 126.9 (1), 126.0 (1), 78.1 (0), 76.4 (1), 75.8 (2), 73.4 (2), 70.7 (2), 68.6 (1), 64.3 (2), 32.2 (2). Anal. Calcd for C<sub>26</sub>H<sub>28</sub>O<sub>4</sub>: C, 77.2; H, 7.0. Found: C, 76.8; H, 7.0.

(4*S*\*,6*R*\*)-4-Benzyloxy-6-benzyloxymethyl-6-methyl-dihydropyran-3-one (6a). To a solution of the alcohol ( $3S^*$ , 4*S*\*, 6*R*\*)-5a (2.12 g, 6.2 mmol) in DCM (20 mL) were added NMO (1.09 g, 9.3 mmol) and powdered 4 Å molecular sieves (3.10 g). TPAP (0.11 g, 5 mol %) was added, and the mixture was stirred at ambient temperature until the starting material was completely consumed as indicated by TLC. The solvent was evaporated, and the residue was filtered through a small pad of silica. The organics were eluted with MTBE, and the solvent was evaporated to give the ketone **6a** (1.76 g, 84%). IR (neat): 699 m, 1101 s, 1725 s. MS m/z (rel intensity): 341 (M<sup>+</sup> + 1, 10), 323 (60), 233 (70), 181 (100), 144 (55), 113 (55). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.29–7.19 (10H), 4.73 (d, 1H, *J* = 11.8 Hz), 4.49 (d, 1H, *J* = 12.3 Hz), 4.46 (d, 1H, *J* = 11.8 Hz), 4.43 (d, 1H, *J* = 12.3 Hz), 4.43 (dd, 1H, *J* = 12.5, 6.5 Hz), 4.13 (d, 1H, *J* = 18.5 Hz), 3.29 (d, 1H, *J* = 9.5 Hz), 2.45 (dd, 1H, *J* = 13.5, 6.5 Hz), 1.67 (dd, 1H, *J* = 13.5, 12.5 Hz), 1.16 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  209.9 (0), 137.8 (0), 137.6 (0), 128.4 (1), 128.3 (1), 127.8 (1), 127.7 (1), 127.6 (1), 76.7 (2), 76.5 (1), 75.6 (0), 73.5 (2), 72.2 (2), 68.6 (2), 36.9 (2), 22.7 (3).

(4S\*,6S\*)-4-Benzyloxy-6-benzyloxymethyl-6-phenyldihy**dropyran-3-one (6b).** Following the procedure for the methyl analogue, we obtained 6b (0.90 g, 82%) from 5b (1.10 g, 2.7 mmol, 4.5:1 mixture of regioisomers). IR (neat): 699 s, 1101 s, 1746 s, 2852 m, 3030 m. MS  $\mathit{m/z}$  (rel intensity): 403 (M^+ + 1, <5), 292 (5), 264 (5), 219 (15), 189 (20), 115 (15), 91 (100), 65 (20). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.33-7.11 (15H), 4.82 (dd, 1H, J = 13.0, 6.3 Hz), 4.79 (d, 1H, J = 11.8 Hz), 4.52 (d, 1H, J = 11.8 Hz), 4.46 (d, 1H, J = 12.3 Hz), 4.34 (d, 1H, J =12.3 Hz), 4.14 (d, 1H, J = 18.0 Hz), 3.95 (d, 1H, J = 18.0 Hz), 3.50 (d, 1H, J = 10.0 Hz), 3.33 (d, 1H, J = 10.0 Hz), 2.99 (dd, 1H, J = 13.0, 6.3 Hz), 2.20 (dd, 1H, J = 13.0, 13.0 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  210.3 (0), 141.9 (0), 137.7 (0), 137.7 (0), 128.5 (1), 128.4 (1), 127.9 (1), 127.8 (1), 127.7 (1), 127.5 (1), 125.6 (1), 79.9 (0), 79.2 (2), 76.7 (1), 73.6 (2), 72.6 (2), 69.3 (2), 35.5 (2). Anal. Calcd for C<sub>26</sub>H<sub>26</sub>O<sub>4</sub>: C, 77.6; H, 6.5. Found: C, 77.3; H, 6.3.

(3R\*,4S\*,6R\*)-4-Benzyloxy-6-benzyloxymethyl-6-methyl-3-vinyl-tetrahydropyran-3-ol (7a). Vinylmagnesium chloride (1.7 M solution in THF, 0.6 mL, 1.0 mmol) was added to a solution of the ketone 6a (0.11 g, 0.3 mmol) in ether (10 mL) at -78 °C. The mixture was stirred until the starting material was fully converted and warmed to ambient temperature, and the reaction was quenched with saturated NH<sub>4</sub>Cl solution. The aqueous layer was extracted with MTBE, and the combined organic extracts were dried with MgSO<sub>4</sub>. Upon flash chromatography on silica, pure 7a (82 mg, 66%) was isolated. IR (neat): 698 s, 1096 s, 1454 m, 2867 m, 3446 br s. MS m/z (rel intensity): 351 (M<sup>+</sup> - 17, 10), 333 (40), 243 (50), 153 (90), 91 (100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.27–7.18 (10H), 5.78 (dd, 1H, J = 17.3, 10.8 Hz), 5.40 (d, 1H, J = 17.3 Hz), 5.14 (d, 1H, J = 10.8 Hz), 4.57 (d, 1H, J = 14.5 Hz), 4.50 (d, 1H, J =11.3 Hz), 4.46 (d, 1H, J = 14.5 Hz), 4.39 (d, 1H, J = 11.3 Hz), 3.59 (dd, 1H, J = 9.0, 4.8 Hz), 3.52 (d, 1H, J = 12.3 Hz), 3.49(d, 1H, J = 12.3 Hz), 3.34 (d, 1H, J = 9.5 Hz), 3.29 (d, 1H, J= 9.5 Hz), 2.61 (bs, 1H), 1.92 (dd, 1H, J = 13.8, 4.8 Hz), 1.72 (dd, 1H, J = 13.8, 9.0 Hz), 1.11 (s, 3H). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  139.5 (1), 138.1 (0), 137.8 (0), 128.4 (1), 127.8 (1), 127.7 (1), 127.6 (1), 127.4 (1), 126.7 (1), 116.0 (2), 76.4 (1), 74.8 (2), 74.2 (0), 73.4 (2), 71.6 (0), 71.3 (2), 67.5 (2), 33.0 (2), 26.9 (3). Anal. Calcd for C23H28O4: C, 75.0; H, 7.7. Found: C, 74.9; H, 7.8

(3R\*,4S\*,6S\*)-4-Benzyloxy-6-benzyloxymethyl-6-phenyl-3-vinyltetrahydropyran-3-ol (7b). Following the procedure for the methyl analogue, we obtained **7b** (0.33 g, 89%) from **6b** (0.35 g, 0.9 mmol). IR (neat): 699 m, 1075 s, 1101 s, 1454 m, 3030 m, 3543 br s. MS m/z (rel intensity): 429 (M<sup>+</sup> + 1, 15), 384 (20), 218 (25), 207 (100), 157 (30), 117 (35), 91 (75), 73 (55). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.38 (d, 2H, J = 7.5Hz), 7.28–7.08 (11H), 6.93 (dd, 2H, J = 7.5, 3.7 Hz), 6.00 (dd, 1H, J = 17.3, 10.8 Hz), 5.48 (dd, 1H, J = 17.3, 1.5 Hz), 5.19 (dd, 1H, J = 10.8, 1.5 Hz), 4.49 (d, 1H, J = 11.3 Hz), 4.43 (d, 1H, J = 12.1 Hz), 4.34 (d, 1H, J = 12.1 Hz), 4.28 (d, 1H, J = 12.1 11.3 Hz), 3.68 (d, 1H, J = 12.0 Hz), 3.64 (dd, 1H, J = 7.3, 4.0 Hz), 3.64 (d, 1H, J = 12.0 Hz), 3.44 (d, 1H, J = 10.3 Hz), 3.40 (d, 1H, J = 10.3 Hz), 2.63 (bs, 1H), 2.40 (dd, 1H, J = 14.3, 4.0 Hz), 2.27 (dd, 1H, J = 14.3, 7.3 Hz). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  143.1 (0), 139.1 (1), 138.0 (1), 137.6 (0), 128.2 (1), 128.2 (1), 128.1 (1), 128.0 (1), 127.7 (1), 127.6 (1), 127.5 (1), 127.5 (1), 127.4 (1), 126.8 (1), 125.3 (1), 116.0 (2), 77.6 (0), 77.3 (1), 76.8 (2), 73.4 (2), 71.2 (2), 70.8 (0), 67.3 (2), 31.1 (2).

**General Procedure for the Preparation of Spirocyclic Lactones.** A solution of the allylic alcohol (3.0 mmol), Rh-(acac)(CO)<sub>2</sub> (1.0 mol %), and BIPHEPHOS (4.0 mol %) in anhydrous dioxane (10 mL) was heated at 60 °C for 20 h under an atmosphere of carbon monoxide (10 bar) and hydrogen (10 bar) in an autoclave. The reaction mixture was filtered through basic alumina, and all organics were eluted with MTBE and then ethanol. The solvent was evaporated, and the crude hemiacetals were redissolved in DCM (20 mL). Powdered molecular sieve (4 Å, 2.00 g) and *N*-methyl morpholine-*N*-oxide (0.53 g, 4.5 mmol) were added, followed by TPAP (53 mg, 5 mol %). The mixture was stirred at ambient temperature until the starting material was completely converted as indicated by TLC (3:1 (v/v) hexanes/ethyl acetate). The solvent was evaporated to one-third of the original volume, and the residue was filtered through a 5 cm pad of silica. The products were eluted with MTBE and purified by flash chromatography, if necessary.

(5R\*,8R\*,10S\*)-10-Benzyloxy-8-benzyloxymethyl-8methyl-1,7-dioxaspiro[4.5]decan-2-one (13a). The title compound was obtained from 7a (0.19 g, 0.5 mmol) following the general procedure for the preparation of spirocyclic lactones. Yield: 0.16 g (80%). IR (neat): 1077 s, 1218 m, 1454 m, 1727 m, 1770 s, 2930 m. <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  7.37-7.24 (10H), 4.67 (d, 1H, J = 12.2 Hz, -CH(4)OCHHPh), 4.53 (d, 1H, J = 12.2 Hz, -CHHOC*H*HPh), 4.50 (d, 1H, J = 12.2Hz, -CHHOCHPh), 4.41 (d, 1H, J = 12.2 Hz, -CH(4)-OCHHPh), 3.73 (d, 1H, J = 12.7 Hz, H6), 3.72 (dd, 1H, J = 10.7, 4.4 Hz, H4), 3.70 (d, 1H, J = 12.2 Hz, H6), 3.50 (d, 1H, J = 9.8 Hz, -CHHOCHHPh), 3.37 (d, 1H, J = 9.8 Hz, -CHHOCHHPh), 2.69 (ddd, 1H, J = 18.0, 10.5, 9.0 Hz, (O)-CCHHCHH-), 2.43 (ddd, 1H, J = 18.0, 10.5, 4.0 Hz, (O)-CCH*H*CHH–), 2.14 (dd, 1H, *J*=13.3, 4.4 Hz, H3eq), 1.92 (ddd, 1H, J = 12.8, 10.5, 4.0 Hz, (O)CCHHCHH-), 1.82 (dd, 1H, J = 13.3, 10.7 Hz, H3ax), 1.72 (ddd, 1H, J = 12.8, 10.5, 9.0 Hz, (O)CCHHCHH-), 1.21 (s, 3H, -CH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.8 (0), 138.0 (0), 137.9 (0), 128.4 (1), 128.4 (1), 127.8 (1), 127.7 (1), 127.6 (1), 127.4 (1), 83.4 (0), 76.1 (2), 75.0 (0), 74.7 (2), 73.5 (2), 70.3 (2), 68.3 (2), 33.5 (2), 28.8 (2), 27.5 (2), 26.1 (2). Anal. Calcd for C24H28O5: C, 72.7; H, 7.1. Found: C, 72.4; H, 7.1.

(5R\*,8R\*,10S\*)-10-Benzyloxy-8-benzyloxymethyl-8phenyl-1,7-dioxaspiro[4.5]decan-2-one ((5R\*,8R\*,10S\*)-13b) and (5*S*\*,6*R*\*,9*S*\*)-6-Benzyloxy-9-benzyloxymethyl-9-phenyl-1,8-dioxaspiro[4.5]decan-2-one ((5S\*,6R\*,9S\*)-13b). The title compounds were obtained from 7b (0.53 g, 1.2 mmol) following the general procedure for the preparation of spirocyclic lactones as a 4:1 mixture of regioisomers that were separated by column chromatography. Data for  $(5R^*, 8R^*, -$ 10S\*)-13b. Yield: 0.14 g (44%). IR (neat): 1075 s, 1775 s, 3033 s. MS m/z (rel intensity): 337 (M<sup>+</sup> – PhCH<sub>2</sub>OCH<sub>2</sub>–, 100%), 229 (32), 91 (C<sub>7</sub>H<sub>7</sub><sup>+</sup>, 98). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz): δ 7.59 (d, 2H, J = 8.0 Hz, Ph), 7.24 (dd, 2H, J = 7.7, 7.7 Hz, Ph), 7.20-7.08 (11H, Ph), 4.39 (d, 1H, J = 12.2 Hz, CH(4)OCH<sub>2</sub>-Ph), 4.22 (d, 1H, J = 12.2 Hz, CH(4)OCH<sub>2</sub>Ph), 4.19 (d, 1H, J = 12.5 Hz,  $-CH_2OCH_2Ph$ ), 4.17 (d, 1H, J = 12.5 Hz,  $-CH_2$ -OCH<sub>2</sub>Ph), 3.77 (d, 1H, J = 12.2 Hz, H6), 3.61 (dd, 1H, J = 9.7, 4.2 Hz, H4), 3.51 (d, 1H, J = 10.0 Hz, H6), 3.50 (d, 1H, J = 12.2 Hz,  $-CH_2OCH_2Ph$ ), 3.40 (d, 1H, J = 12.2 Hz,  $-CH_2OCH_2$ -Ph), 2.44 (dd, 1H, J = 13.5, 4.2 Hz, H3eq), 2.33 (ddd, 1H, J = 17.5, 10.5, 8.2 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.26 (dd, 1H, J = 13.5, 9.7 Hz, H3ax), 1.94 (ddd, 1H, J = 17.5, 10.7, 4.7 Hz, (O)CCH<sub>2</sub>-CH<sub>2</sub>-), 1.37 (ddd, 1H, J = 13.0, 10.5, 4.7 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.20 (ddd, 1H, J = 13.0, 10.5, 8.2 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.7 (0), 143.7 (0), 137.9 (0), 137.8 (0), 128.4 (1), 128.3 (1), 128.0 (1), 127.7 (1), 127.6 (1), 127.5 (1), 127.4 (1), 127.1 (1), 125.1 (1), 83.4 (0), 78.2 (0), 76.4 (1), 75.6 (2), 73.5 (2), 70.6 (2), 67.9 (2), 33.2 (2), 28.7 (2), 27.8 (2). Data for (5*S*\*,6*R*\*,9*S*\*)-13b. Yield: 0.03 g (10%). IR (neat): 1096 s, 1454 m, 1778 s, 2928 s. MS m/z (rel intensity): 336 (M<sup>+</sup> PhCH<sub>2</sub>OCH<sub>3</sub>, 86%), 91 (C<sub>7</sub>H<sub>7</sub><sup>+</sup>, 100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  7.58 (d, 2H, J = 8.6 Hz, Ph), 7.41–7.06 (13H, Ph), 4.48 (d, 1H, J = 12.3 Hz,  $-CH_2OCH_2Ph$ ), 4.35 (d, 1H, J = 12.3Hz,  $-CH_2OCH_2Ph$ ), 4.23 (d, 1H, J = 12.3 Hz,  $-CH(4)OCH_2$ -Ph), 4.03 (d, 1H, J = 12.3 Hz, -CH(4)OCH<sub>2</sub>Ph), 3.80 (dd, 1H, J = 10.8, 10.8 Hz, H6ax), 3.71 (dd, 1H, J = 10.8, 4.8 Hz, H6eq), 3.51 (d, 1H, J = 10.0 Hz,  $-CH_2OCH_2Ph$ ), 3.32 (d, 1H, J = 10.0Hz,  $-CH_2OCH_2Ph$ ), 3.24 (dd, 1H, J = 10.8, 4.8 Hz, H5), 2.49 (d, 1H, J = 15.0 Hz, H3eq), 2.43 (ddd, 1H, J = 17.6, 10.0, 10.0

Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.31 (d, 1H, J = 15.0 Hz, H3ax), 2.05 (ddd, 1H, J = 17.6, 10.8, 3.3 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.69 (ddd, 1H, J =13.1, 10.5, 3.3 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.36 (ddd, 1H, J = 13.1, 10.8, 10.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$ 175.8 (0), 141.6 (0), 139.6 (0), 139.0 (0), 129.2 (1), 129.1 (1), 128.9 (1), 128.5 (1), 128.4 (1), 128.3 (1), 128.2 (1), 128.1 (1), 127.8 (1), 83.4 (0), 80.3 (2), 79.9 (0), 79.4 (1), 74.4 (2), 72.2 (2), 60.9 (2), 39.8 (2), 33.1 (2), 29.7 (2).

(5.S,6.S)-6-Methyl-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5.S, 6S)-15a) and (5R,6S)-6-Methyl-1,7-dioxaspiro[4,5]dec-9en-2-one ((5R,6S)-15a). The title compounds were obtained from 9a (3:1 mixture of diastereoisomers, 1.3 g, 9.3 mmol) as a 3:1 mixture of diastereoisomers. Separation of the diastereoisomers by column chromatography on silica gives (5S, 6S)15a (0.95 g, 61%) and (5R,6S)-15a (0.31 g, 20%). Analytical data for (5*S*, 6*S*)-15a. Mp = 103 °C. IR (KBr, disk): 954 s, 1132 s, 1377 s, 1469 s, 1767 s, 2998 s. MS *m*/*z* (rel intensity): 168 (M<sup>+</sup>, 100), 108 (20). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  6.02 (ddd, 1H, J = 10.0, 3.3, 1.8 Hz, H5), 5.74 (ddd, 1H, J = 10.0, J2.0, 2.0 Hz, H4), 4.22 (ddd, 1H, J = 17.1, 3.3, 2.0 Hz, H6eq), 4.09 (ddd, 1H, J = 17.1, 2.0, 2.0 Hz, H6ax), 3.62 (q, 1H, J = 6.5 Hz, H2), 2.64 (ddd, 1H, J = 18.2, 10.5, 7.5 Hz, (O)CCH<sub>2</sub>-CH<sub>2</sub>-), 2.56 (ddd, 1H, J = 18.2, 9.8, 6.5 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.12 (m, 1H, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.01 (m, 1H, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.24 (d, 3H, J = 6.5 Hz,  $-CH_3$ ).<sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.2 (0), 130.7 (1), 126.4 (1), 80.9 (0), 75.4 (1), 64.1 (2), 29.9 (2),28.0 (2), 13.9 (3).  $[\alpha]^{20}_{D} = +171^{\circ}$  (*c* 1.55, CH<sub>2</sub>Cl<sub>2</sub>). Anal. Calcd for C<sub>9</sub>H<sub>12</sub>O<sub>3</sub>: C, 64.3; H, 7.2. Found: C, 64.3; H, 7.2. Analytical data for (5R,6S)-15a. Mp = 102 °C. IR (KBr, disk): 727 m, 959 s, 1069 s, 1227 s, 1759 s, 2999 s. MS *m*/*z* (rel intensity): 168 (M<sup>+</sup>, 100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz):  $\delta$  5.81 (ddd, 1H, J = 10.3, 2.5, 1.3 Hz), 5.77 (ddd, 1H, J = 10.3 Hz), 4.16 (dm, 1H, J = 16.8 Hz), 4.06 (dm, 1H, J = 16.8 Hz), 3.71 (q, 1H, J = 6.5 Hz), 2.56 (ddd, 1H, J = 18.4, 10.0, 4.8 Hz), 2.47 (ddd, 1H, J = 18.4, 10.0, 8.6 Hz), 2.37 (ddd, 1H, J = 13.3, 10.0, 4.8 Hz), 1.93 (ddd, 1H, J = 13.3, 10.0, 8.6 Hz), 1.14 (d, 3H, J = 6.5 Hz, -CH<sub>3</sub>). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz): δ 176.7 (0), 129.5 (1), 128.2 (1), 82.5 (0), 73.8 (1), 65.6 (2), 29.2 (2), 27.1 (2), 14.4 (3).  $[\alpha]^{20}_{D} = +161^{\circ}$  (*c* 1.60, CHCl<sub>3</sub>). Anal. Calcd for C<sub>9</sub>H<sub>12</sub>O<sub>3</sub>: C, 64.3; H, 7.2. Found: C, 64.2; H, 7.1.

(5S\*,6S\*)-6-Phenyl-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5S\*,6S\*)-15b) and (5R,6S)-6-Phenyl-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5R\*,6S\*)-15b). The title compounds were obtained from 9b (0.60 g, 3.0 mmol) as a 4:1 mixture of diastereoisomers. Separation of the diastereoisomers by column chromatography on silica gives (5.S,6.S)-15b (0.26 g, 38%) and (5R\*,6S\*)-15b (0.10 g, 14%). Analytical data for (5S\*,6S\*)-**15b**. Mp = 164 °C. IR (KBr, disk): 708 s, 1032 s, 1201 s, 1456 m, 1767 s, 2831 m, 2945 m. MS *m*/*z* (rel intensity): 231 (M<sup>+</sup> + 1, 20), 157 (45), 141 (100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz): δ 7.44 (d, 2H, J = 7.3 Hz, ortho-H, Ph), 7.37-7.32 (3H, Ph), 6.16 (ddd, 1H, J = 10.3, 3.5, 1.5 Hz, H5), 5.84 (ddd, 1H, J = 10.3, 2.0, 2.0 Hz, H4), 4.46 (ddd, 1H, J = 17.2, 3.4, 2.0 Hz, H6eq), 4.42 (s, 1H, H2), 4.32 (ddd, 1H, J = 17.2, 1.5, 1.5 Hz, H6ax), 2.24-2.16 (2H, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.01 (m, 1H, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.49 (m, 1H, (O)CCH<sub>2</sub>CH<sub>2</sub>-). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.2 (0), 135.4 (0), 131.9 (1), 128.9 (1), 128.7 (1), 128.4 (1), 127.5 (1), 83.1 (1), 80.3 (0), 66.1 (2), 30.0 (2), 27.9 (2). Anal. Calcd for C<sub>14</sub>H<sub>14</sub>O<sub>3</sub>: C, 73.0; H, 6.1. Found: C, 73.0; H, 6.2. Analytical data for  $(5R^*, 6S^*)$ -15b. Mp = 110 °C. IR (KBr, disk): 704 m, 1028 s, 1092 s, 1454 s, 1692 m, 1778 s, 2941 s. MS m/z (rel intensity): 231 (M<sup>+</sup> + 1, 100), 124 (55), 96 (47). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  7.41 (d, 2H, J = 7.3 Hz, ortho-H, Ph), 7.37-7.29 (3H, Ph), 5.95 (ddd, 1H, J=10.3, 2.5, 2.0 Hz, H4,5), 5.90 (ddd, 1H, J = 10.3, 2.0, 2.0 Hz, H4,5), 4.67 (s, 1H, H2), 4.40-4.33 (2H, H6), 2.33 (ddd, 1H, J=13.7, 10.7, 2.4 Hz, (O)-CCH<sub>2</sub>CH<sub>2</sub>-), 2.03 (ddd, 1H, J = 18.1, 10.7, 2.4 Hz, (O)CCH<sub>2</sub>-CH<sub>2</sub>-), 1.86 (ddd, 1H, J = 13.7, 10.5, 10.5 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 0.92 (ddd, 1H, J = 18.1, 10.5, 10.5 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz): δ 177.0 (0), 136.0 (0), 130.0 (1), 128.5 (1), 128.3 (1), 128.0 (1), 126.6 (1), 82.5 (0), 78.5 (1), 66.3 (2), 28.4 (2), 27.9 (2). Anal. Calcd for  $C_{14}H_{14}O_3{:}\ C,\ 73.0;\ H,\ 6.1.$ Found: C, 72.8; H, 6.1.

(5*R*\*,8*S*\*)-8-(4-Methoxyphenyl)-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5*R*\*,8*S*\*)-17). The title compound was obtained from (3R\*,6S\*)-11 (0.50 g, 2.2 mmol) as a colorless solid (0.33 g, 58%). Purification was achieved by flash chromatography on silica and recrystallization from DCM/hexane. Mp = 64 °C. IR (KBr, disk): 1032 s, 1244 s, 1304 s, 1514 s, 1612 s, 1775 s, 2838 m, 2959 s. MS m/z (rel intensity): 260 (M<sup>+</sup>, 76), 162 (61), 135 (100). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  7.21 (d, 2H, J = 8.3Hz, ortho-H, Ar), 6.87 (d, 2H, J = 8.3 Hz, meta-H, Ar), 5.96 (d, 1H, J = 10.7 Hz, H3,4), 5.93 (d, 1H, J = 10.7 Hz, H3,4), 5.10 (s, 1H, H2), 3.86 (d, 1H, J = 11.2 Hz, H6eq), 3.79 (d, 1H, J = 11.2, H6ax), 3.78 (s, 3H, OMe), 2.66–2.56 (2H, (O)CC $H_2$ -CH<sub>2</sub>-), 2.45 (ddd, 1H, J = 13.2, 8.3, 5.4 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.18 (ddd, 1H, J = 13.2, 9.8, 9.3 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-). <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.2 (0), 159.7 (0), 132.6 (1), 131.4 (0), 128.7 (1), 128.0 (1), 114.0 (1), 79.3 (0), 76.0 (1), 69.2 (2), 55.3 (3), 31.3 (2), 27.8 (2). Anal. Calcd for  $C_{15}H_{16}O_4$ : C, 69.2; H, 6.2. Found: C, 68.9; H, 6.2.

(5*S*\*,8*S*\*)-8-(4-Methoxyphenyl)-1,7-dioxaspiro[4,5]dec-9-en-2-one ((5*R*\*,8*S*\*)-17). The title compound was obtained from (5*S*\*,8*S*\*)-17 (0.80 g, 3.4 mmol) as a colorless solid (0.48 g, 54%). Purification was achieved by flash chromatography on silica and recrystallization from DCM/hexane. Mp = 218 °C. IR (KBr, disk): 967 s, 1173 s, 1437 m, 1594 m, 1774 s, 2960 m. MS m/z (rel intensity): 260 (M<sup>+</sup>, 100), 229 (28), 162 (45), 135 (90). <sup>1</sup>H NMR (CDCl<sub>3</sub>, 600 MHz):  $\delta$  7.29 (d, 2H, J = 8.5 Hz, ortho-H, Ar), 6.87 (d, 2H, J = 8.5 Hz, meta-H, Ar), 6.03 (d, 1H, J = 10.3, 2.0 Hz, H3), 5.90 (d, 1H, J = 10.3 Hz, H4), 5.03 (s, 1H, H2), 3.97 (d, 1H, J = 11.7 Hz, H6eq), 3.77 (s, 3H, OMe), 3.66 (d, 1H, J = 11.7, H6ax), 2.66 (ddd, 1H, J = 18.1, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.19 (ddd, 1H, J = 18.1, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.15 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 2.10 (ddd, 1H, J = 13.2, 10.0, 7.0 Hz, (O)CCH<sub>2</sub>CH<sub>2</sub>-), 1.3C NMR (CDCl<sub>3</sub>, 100 MHz):  $\delta$  176.0 (0), 159.6 (0), 134.0 (1), 131.0 (0), 129.0 (1), 126.0 (1), 113.8 (1), 78.9 (0), 75.9 (1), 69.6 (2), 55.2 (3), 30.1 (2), 28.0 (2). Anal. Calcd for C<sub>15</sub>H<sub>16</sub>O<sub>4</sub>: C, 69.2; H, 6.2. Found: C, 69.0; H, 6.0.

**Acknowledgment.** We thank the Deutsche Forschungsgemeinschaft and the Fonds der Chemischen Industrie for generous financial support.

JO0157730